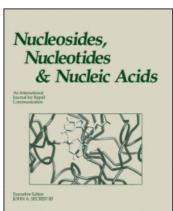
This article was downloaded by:

On: 27 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



# Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713597286

# The tRNA "WOBBLE POSITION" Uridines. III.¹ the Synthesis of 5-[S-Methoxycarbonyl (Hydroxy)Methyl] Uridine and its 2-Thio Analogue

Barbara Nawrot<sup>a</sup>; Andrzej Malkiewicz<sup>a</sup>

<sup>a</sup> Institute of Organic Chemistry, Technical University (Politechnika), Łódz. zwirki 36, Poland

To cite this Article Nawrot, Barbara and Malkiewicz, Andrzej (1989) 'The tRNA "WOBBLE POSITION" Uridines. III. the Synthesis of 5-[S-Methoxycarbonyl (Hydroxy)Methyl] Uridine and its 2-Thio Analogue', Nucleosides, Nucleotides and Nucleic Acids, 8: 8, 1499-1512

To link to this Article: DOI: 10.1080/07328318908048857 URL: http://dx.doi.org/10.1080/07328318908048857

# PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

THE tRNA "WOBBLE POSITION" URIDINES. III. 1
THE SYNTHESIS OF 5-[S-METHOXYCARBONYL(HYDROXY)METHYL]
URIDINE AND ITS 2-THIO ANALOGUE.

### Barbara Nawrot and Andrzej Małkiewicz

Institute of Organic Chemistry, Technical University (Politechnika), 90-924 Łódź, Zwirki 36, Poland

Abstract. The diastereoisomers  $\underline{2a}$ ,  $\underline{2b}$  and their 2-thio analogues  $\underline{4a}$  and  $\underline{4b}$  were obtained by three-step transformation of uridine and 2-thiouridine, respectively. The absolute configuration at C-5<sup>1</sup> in  $\underline{2a}$  and  $\underline{2b}$  was established by CD, while for  $\underline{4a}$  and  $\underline{4b}$  the configurational assignment was based on the chemical correlation. The acids  $\underline{1}$  and  $\underline{3}$  were obtained by alkaline hydrolysis of  $\underline{2a}$  and  $\underline{4a}$ , respectively.

#### Introduction

An ability of  $tRNA_1^{Gly}$  (Bombyx mori posterior silk gland) as well as tRNA<sub>2</sub>GlY (of the same origin) for decoding within the glycine family codons is controversial  $^{2,3}$ . Furthermore. either 5-[S-carboxy(hydroxy)methyl]uridine  $(1, chm^5U)$  or 5[S-methoxycarbonyl(hydroxy)methyl]uridine (<math>2a, mchm<sup>5</sup>U) have been identified in tRNA<sub>2</sub>Gly "wobble position"  $(SCHEME 1)^{4,5}$ . One can argue that at least three isoacceptor tRNAs are necessary to translate certain synonymous codons contexts on the fibroin mRNA. Indeed, in the case of specialized tissues, which produce predominantly one protein, the relationship between the frequency of synonymous codon usage and abundance of isoacceptor tRNAs has been clearly demonstrated<sup>6,7,8</sup>. However, the influence of the "wobble position" uridines on the mRNA-ribosome-tRNA complex binding energy and selection on this way from among competing tRNA species remains not clear so  $far^{8,9}$ . It seems that steric puckering of the modified uridines modulates dynamics of the "extended anticodon" region conformation  $^{10-14}$ .

### SCHEME 1

In continuation of our work on the preparation of models for the study of the conformation of tRNA modified units  $^{14}$  and anticodon loop fragments  $^{15}$  we present the synthesis of  $\frac{1}{2}$  and  $\frac{2a}{2}$  as well as  $\frac{3}{2}$  and  $\frac{4a}{2}$ , which can be considered as potential components of the biopolymer sequences  $^{9}$ .

### Results

Kawakami et al.<sup>5</sup> have briefly reported a seven-step transformation of 5-bromo-2,4-di-tert-butoxypyrimidine to  $\underline{1}$  and  $\underline{2}$ . However, neither the synthetic procedure nor spectroscopic and chromatographic data have been as yet described.

The recently reported method for the uridine regioselective 5-lithiation, controlled by tert-butyldimethylsilyl protection of the sugar moiety $^{16}$ , allows us to simplify substantially the above mentioned approach.

Thus, treatment of 2',3',5'-tris-0-(tert-butyldimethylsilyl)uridine  $(\underline{5})^{17}$  with n-butyl lithium in tetrahydrofuran

in the presence of tetramethylenediamine (TMEDA) formed dianion  $\underline{7}$  which was then condensed at  $-78^{\circ}\text{C}$  with freshly distilled n-butyl glyoxylate  $\underline{18}$  to give  $\underline{9a}$  and  $\underline{9b}$  (SCHEME 2). Diastereoisomers  $\underline{9a}$  and  $\underline{9b}$  were separated from the reaction mixture in the ratio 2:1 by column chromatography on silica gel in ca. 30% yield.

# SCHEME 2

$$\begin{array}{c|c}
\bullet & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline
SiO & \bullet & \bullet & \bullet \\
\hline$$

<u>9a</u>

RT, 24h

An analogous alkylation of 2',3',5'-tris-O-(tert-butyl-dimethylsilyl)-2-thiouridine (6) underwent in higher yield (40%) leading to diastereoisomers 10a and 10b in the ratio 1:1. The spectral data of both pairs of diastereoisomers 9a, 9b and 10a, 10b fully confirm the 5-substitution of persilylated uridine or 2-thiouridine with n-butoxycarbonyl (hydroxy)methyl group.

Treatment of individual diastereoisomers  $\underline{9a}$ ,  $\underline{9b}$ ,  $\underline{10a}$  and  $\underline{10b}$  with methanolic hydrogen chloride leads to the removal of tert-butyldimethylsilyl groups with simultaneous transesterification giving the title nucleosides  $\underline{2a}$ ,  $\underline{2b}$ ,  $\underline{4a}$  and  $\underline{4b}$ , respectively.

Separation of 2-oxo and 2-thio diastereoisomeric pairs is best accomplished with protected nucleosides for which more distinctive differences of  $R_f$  values were noticed (TABLE 1). Deprotected nucleosides  $\underline{2a}$ ,  $\underline{2b}$ ,  $\underline{4a}$  and  $\underline{4b}$  show significantly different HPLC retention times and can be identified by this technique (TABLE 1).

The  $^1$ H NMR spectra of  $^2$ 2a,  $^2$ 2b,  $^4$ 2a and  $^4$ 2b have no diagnostic value for the determination of the absolute configuration at  $^2$ 2-1 atom or for the identification of diastereoisomers within 2-oxo or 2-thio pairs, e.g.  $^1$ H NMR spectrum of the equimolar mixture of  $^2$ 2a and  $^2$ 2b exhibits only minute differences of the chemical shifts for the corresponding H-6 and H-1' protons ( $^2$ 6Hz,  $^2$ 6Hz,  $^2$ 6Hz,  $^2$ 71.2Hz at 300 MHz).

The absolute configuration at the asymmetric  $C-5^1$  atom of 2a and 2b was established by comparison of their CD spectra with those of substituted S(+) methyl mandelates as well as S(+) methyl atrolactate S(+). Model compounds show a weak negative band at ca. 260 nm, which is assigned to the aromatic S(+) band, and a second, much stronger, positive band at ca. 220 nm, which is possibly due to the interaction between the carbonyl and aromatic S(+) transitions.

Both nucleosides  $\underline{2a}$  and  $\underline{2b}$  show CD spectra exhibiting positive Cotton effects within 260-280 nm and 195-210 nm spectral regions. CD maxima of the opposite signs ( $\underline{2a}$ : $\Delta \in$ 

TABLE	1.	TLC and HPLC chromatographic data of the synthe-
		sized nucleosides

Nucleoside	TLC silica gel R <sub>f</sub> (solvent system)*	TLC cellulose R <sub>f</sub> (solvent system)*	HPLC* Retention time (min)
		0.18(E)	6.62 <sup>c</sup>
1	<del></del>		8.01 <sup>b</sup>
<u>3</u>	<del>-</del>	0.19(E)	
<u>2a</u>	0.13(C)	0.73(E)	3.50ª
<u>2b</u>	0.11(C)	0.73(E)	5.00 <sup>a</sup>
<u>4a</u>	0.34(C)	0.38(F)	10.00ª
<u>4b</u>	0.27(C)	0.38(F)	16.50ª
<u>9a</u>	0.11(A)	-	-
<u>9b</u>	0.17(A)	-	-
<u>10a</u>	0.32(A)	-	-
<u>10b</u>	0.47(A)	-	_

<sup>\*</sup> Solvent systems and HPLC conditions are described in Experimental part.

+1.06,  $2b:\Delta \in -2.28$ ), however, are observed for the 220 nm transition (FIGURE 1). In conclusion chirality of C-5<sup>1</sup> of 2a is proposed to be S. This observation is further supported by the ORD spectra of nucleosides 2a and 2b (FIGURE 1) when compared with those obtained by Kawakami et al.<sup>5</sup> for the analogues of 2a and 2b containing reduced side chain.

CD spectra of 2-thio derivatives  $\underline{4a}$  and  $\underline{4b}$  are almost identical with the spectrum of 2-thiouridine<sup>21</sup> in the 240-360 nm range, but they exhibit additional negative band at ca. 220 nm, which is "less negative" for  $\underline{4a}$  (FIGURE 1). Although it may suggest the S configuration at C-5<sup>1</sup> in  $\underline{4a}$ , we found a more reliable proof of the configuration of 2-thio analogues which is based on the chemical correlation. Thus, under the non-racemizing conditions ( $\mathrm{H}_2\mathrm{O}_2$ , pH 8.0, 20

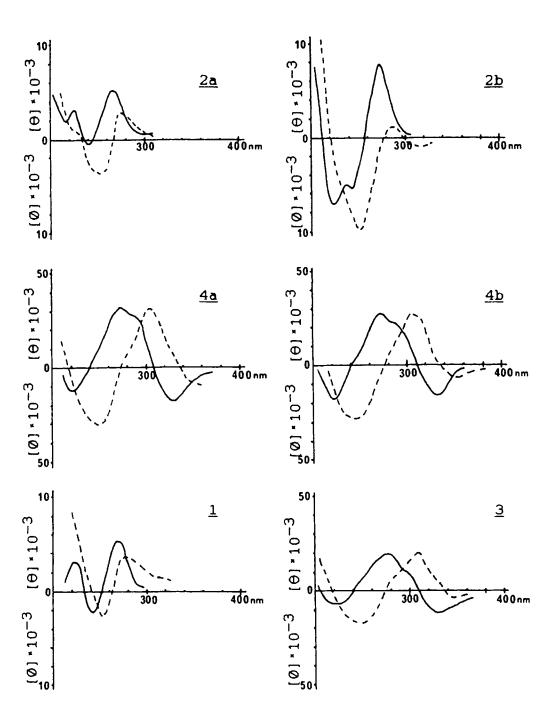


FIGURE 1. CD (\_\_\_\_\_) and ORD (----) spectra of modified uridines and their 2-thio analogues.

h)  $^{22}$   $\underline{_{4a}}$  and  $\underline{_{4b}}$  were transformed exclusively to  $\underline{_{2a}}$  and  $\underline{_{2b}}$ , respectively, as it was demonstrated by the HPLC analysis of the crude products (SCHEME 3). Consequently,  $\underline{_{2a}}$  and  $\underline{_{4a}}$  have the S absolute configuration at C-5<sup>1</sup>, while  $\underline{_{2b}}$  and  $\underline{_{4b}}$  are their C-5<sup>1</sup> epimers.

### SCHEME 3

Finally, diastereoisomeric esters  $\underline{2a}$  and  $\underline{4a}$  were hydrolyzed under non-racemizing conditions  $^{5,23}$  (0.01N NaOH/H<sub>2</sub>O, RT, 48h) into the nucleosides  $\underline{1}$  and  $\underline{3}$ , respectively. The purification of the free acids was accomplished by paper chromatography. Compounds  $\underline{1}$  and  $\underline{3}$  were characterized by  $^{1}$ H NMR, CD and ORD (FIGURE 1), UV and HPLC (TABLE 1). Comparison of our spectral data for  $\underline{1}$  with those reported earlier by Kawakami et al. for the natural product  $^{4,5}$  shows their identity.

# Experimental

 $^{1}\text{H}$  NMR spectra: TESLA BS 467 (60 MHz) - TMS was used as the internal standard and BRUKER MSL-300 (300 MHz) - TMS was

used as the external standard. Electron impact mass spectra (MS) at 70 eV: GS MS LKB 2091 instrument and field desorption mass spectra (FD MS) at 15 eV: Varian-Mat 711.

CD and ORD spectra: JASCO J-20 Automatic Recording Spectropolarimeter, room temperature, 10 mm cell.

High-performance liquid chromatography (HPLC) was carried out on Laboratorni Pristroje (Praha) equipped with a UV VIS Detector LCD 2563. Separon SGX  $C_{18}$  5  $\mu m$  glass column (150 x 3.3 mm), 2.5% acetonitrile in water, flow rates: 1 mL/min (a), 0.6 mL/min (b). LiChrosorb RP-2 (Merck) 5  $\mu m$  stainless steel column (250 x 4.6 mm), 5% acetonitrile in water, flow rate 0.66 mL/min (c).

Thin layer chromatography (TLC) was performed on silicagel 60  $F_{254}$  (Merck) plates in the solvent systems (v/v): A: chloroform/methanol 97/3, B: ethyl ether/hexane 2/1, C: chloroform/methanol 85/15 or on cellulose F (Merck) plates: E: isopropanol/water 7/3, F: n-butanol/water (saturated solution).

Silica gel  $F_{60}$  (230-400 mesh) (Merck) was used for column chromatography and 3MM Whatman paper was applied for paper chromatography.

Evaporations were carried out under a reduced pressure and bath temperature below 40°C.

### 2',3',5'-tris-O-(tert-butyldimethylsilyl)-2-thiouridine (6)

To the solution of 2-thiouridine (2 mmol, 520 mg) and imidazole (28 mmol, 1.90 g) in anhydrous DMF (20 mL) tertbutyldimethylsilyl chloride (14 mmol, 2.11 g) was added. The reaction was carried out at room temperature for 24 h and the solvent was evaporated under reduced pressure. The residue was co-evaporated with n-butanol, toluene and purified on silica gel short column (solvent system B) to give  $\underline{6}$  as a foam (1.08 g, 90% yield).  $R_f = 0.82$  (B);  $^1\text{H}$  NMR CDCl<sub>3</sub> ( $\delta$ , ppm): 11.23 (br s., 1H, NH), 8.36 (d,  $J_{5,6} = 8$  Hz, 1H, H-6), 6.35 (d,  $J_{1',2'} = 2$  Hz, 1H, H-1'), 5.93 (d,  $J_{5,6} = 8$  Hz, 1H, H-5), 4.23-3.56 (m, 5H, H-2',3',4',5',5"), 0.90 (m, 45H, TBDMS-protons); MS m/z (%): 587 (1.64, M<sup>+</sup>·-15), 545 (100.00, M<sup>+</sup>·-57), 474 (22.53), 261 (21.83), 185 (32.62).

# 5-[n-butoxycarbonyl(hydroxy)methyl]-2',3',5'-tris-0-(tert-butyldimethylsilyl)uridine (9a) and (9b)

To the solution of 2',3',5'-tris-0-(tert-butyldimethylsilyl)uridine  $(5)^{17}$  (1 mmol, 586.2 mg) and TMEDA (2.5 mmol, 377 μL) in anhydrous THF (10 mL) cooled down to -78°C n-butyl lithium in hexane (2.5 mmol) was added atmosphere). The solution was stirred for 1.5 h and freshly distilled n-butyl glyoxylate  $^{18}$  (2 mmol, 240  $\mu$ L) in THF (2 mL) was added at -78°C. The mixture was stirred for 1 h at -78°C, then warmed to room temperature and kept overnight. Acetic acid (2.5 mmol, 142.5  $\mu$ L) was added and the solvent removed under reduced pressure. The residue was dissolved in chloroform (20 mL), washed with cold 2% sodium bicarbonate and dried  $(MgSO_A)$ . After evaporation of solvent the residue was chromatographed on the silica gel column (gradient of chloroform/methanol 100% to 99% v/v) to diastereoisomers 9a and 9b as foams. 9a (150 mg, 21.6% yield):  $R_f = 0.11$  (A),  ${}^{1}H$  NMR CDCl<sub>3</sub> ( $\delta$ , ppm): 9.30 (br s, 1H, NH), 7.66 (s, 1H, H-6), 5.93 (d,  $J_{1',2'} = 5 \text{ Hz}$ , H-1'), 4.61 (s, 1H,  $H-C-5^1$ ), 4.20-3.56 H-2',3',4',5',5'',  $CH_2OCO)$ , 1.60-1.03 (m, 7H,  $CH_2CH_2CH_3$ ), 0.76 (m, 45H, TBDMS-protons); 9b (76 mg, 10.6% yield):  $R_f =$ 0.17 (A),  ${}^{1}H$  NMR CDCl<sub>3</sub> ( $\delta$ , ppm): 7.85 (s, 1H, H-6), 6.03 (d,  $J_{1',2'} = 5 \text{ Hz}, 1H, H-1'), 4.81 (s, 1H, H-C-5^1), 4.35-3.76$ (m, 7H, H-2', 3', 4', 5', 5'',  $CH_2OCO$ ), 1.65-1.08 (m, 7H,  $CH_2CH_2CH_3$ ), 0.90 (m, 45H, TBDMS-protons).

# 5-[n-butoxycarbonyl(hydroxy)methyl]-2',3',5'-tris-0-(tert-butyldimethylsilyl)-2-thiouridine (10a) and (10b)

Following the procedure for the synthesis of  $\underline{9a}$  and  $\underline{9b}$  2',3',5'-tris-O-(tert-butyldimethylsilyl)-2-thiouridine ( $\underline{6}$ ) (1 mmol, 602 mg) was transformed to the nucleosides  $\underline{10a}$  and  $\underline{10b}$ .  $\underline{10a}$  (155 mg, 21% yield): R<sub>f</sub> = 0.32 (A), <sup>1</sup>H NMR CDCl<sub>3</sub> (6, ppm): 10.06 (br s, 1H, NH), 7.90 (s, 1H, H-6), 6.86 (d, J<sub>1',2'</sub> = 5 Hz, 1H, H-1'), 4.73 (br s, 1H, H-C-5<sup>1</sup>), 4.30-3.53 (m, 7H, H-2',3',4',5',5", CH<sub>2</sub>OCO), 1.63-1.03 (m, 7H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 0.86 (br s, 45H, TBDMS-protons);  $\underline{10b}$  (131 mg,

18% yield)  $R_f = 0.47$  (A),  $^1\text{H}$  NMR CDCl<sub>3</sub> ( $\delta$ , ppm): 9.70 (br s, 1H, NH), 7.93 (s, 1H, H-6), 6.73 (d,  $J_{1',2'} = 5$  Hz, 1H, H-1'), 4.70 (br s, 1H, H-C-5<sup>1</sup>), 4.26-3.50 (m, 7H, H-2',3',4',5',5", CH<sub>2</sub>OCO), 1.58-1.00 (m, 7H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 0.81 (br s, 45H, TBDMS-protons).

# 5-[S-methoxycarbonyl(hydroxy)methyl]uridine (2a)

The persitylated ester ga (0.238 mmol, 171 mg) was dissolved in 1N HCl/MeOH (3 mL) and kept overnight at room temperature. The solvent was evaporated and the residue several times co-evaporated with methanol. Crude product was finally purified on Whatman 3MM paper (solvent system E) to give ga (75 mg, 95% yield) as a foam. ga (0.13 (C), ga = 0.73 (E), ga 1H NMR CD3OD (6, ppm): 8.13 (s, 1H, H-6), 5.92 (d, ga - 4.6 Hz, 1H, H-1'), 4.98 (s, 1H, H-C-5\frac{1}{2}), 4.22-4.14 (m, 2H, H-2',3'), 4.02 (dt, ga - 4.3 Hz, ga - 4.3 Hz, ga - 4.2 Hz, 1H, H-4'), 3.85 (dd, ga - 4.3 Hz, ga - 4.3 Hz, ga - 4.5 Hz, 1H, H-5'), 3.75 (dd, ga - 5" = 3.0 Hz, ga - 5" = 12.2 Hz, 1H, H-5"), 3.73 (s, 3H, CH3); UV (nm) pH 7 (H2O) ga - 267; HPLC ga - 3.5 min (a); FD MS m/z: 273 (M\frac{1}{2}\dots-59).

### 5-[R-methoxycarbonyl(hydroxy)methyl]uridine (2b)

The nucleoside 9b (0.15 mmol, 109 mg) was treated with 1N HCl/MeOH (2.5 mL) according to procedure described for 2a, to give 2b (48 mg, 95% yield).  $R_f = 0.11$  (C),  $R_f = 0.73$  (E),  $^1$ H NMR CD<sub>3</sub>OD ( $\delta$ , ppm): 8.06 (s, 1H, H-6), 5.81 (d,  $J_{1',2'} = 4.^2$  Hz, 1H, H-1'), 4.87 (s, 1H, H-C-5 $^1$ ), 4.12-4.04 (m, 2H, H-2',3'), 3.92 (dt,  $J_{4',5'} = 2.9$  Hz,  $J_{3',4'} = 4.1$  Hz, 1H, H-4'), 3.76 (dd,  $J_{4',5'} = 2.7$  Hz,  $J_{5',5''} = 12.1$  Hz, 1H, H-5'), 3.64 (dd,  $J_{4',5''} = 3.0$  Hz,  $J_{5',5''} = 12.1$  Hz, 1H, H-5''), 3.63 (s, 3H, CH<sub>3</sub>); UV (nm) pH 7 (H<sub>2</sub>O):  $\lambda_{max} = 266$  ( $\epsilon = 5200$ ),  $\lambda_{min} = 235$  ( $\epsilon = 570$ ), pH 12 (NaOH):  $\lambda_{max} = 266$ ; HPLC  $\lambda_{ret} = 5.0$  min (a); FD MS m/z: 333.3 (M<sup>+</sup>+1).

### 5-[S-methoxycarbonyl(hydroxy)methyl]-2-thiouridine (4a)

The compound  $\underline{10a}$  (0.34 mmol, 250 mg) was transformed to the title nucleoside  $\underline{4a}$  using the same procedure as

described for the preparation of 2a. 4a (111 mg, 94% yield):  $R_f = 0.34$  (C),  $R_f = 0.38$  (F),  $^1H$  NMR CD<sub>3</sub>OD ( $\delta$ , ppm): 8.66 (s. 1H, H-6), 6.79 (d.  $J_{1',2'} = 2.5$  Hz, 1H, H-1'), 5.15 (s. 1H, H-C-5<sup>1</sup>), 4.41 (dd.  $J_{1',2'} = 2.5$  Hz,  $J_{2',3'} = 4.8$  Hz, 1H, H-2'), 4.33 (dd.  $J_{2',3'} = 4.8$  Hz,  $J_{3',4'} = 7.0$  Hz, 1H, H-3'), 4.26 (dt.  $J_{3',4'} = 7.0$  Hz,  $J_{4',5'} = 2.4$  Hz, 1H, H-4'), 4.14 (dd.  $J_{4',5'} = 2.4$  Hz,  $J_{5',5''} = 12.4$  Hz, 1H, H-5'), 3.98 (dd.  $J_{4',5''} = 2.61$  Hz,  $J_{5',5''} = 12.4$  Hz, 1H, H-5''), 3.92 (s. 3H, CH<sub>3</sub>), UV (nm) pH 7 (H<sub>2</sub>O):  $\lambda_{max} = 282$  ( $\epsilon = 10250$ ), pH 12 (NaOH):  $\lambda_{max} = 242$ ,  $\lambda_{max} = 276$ ,  $\lambda_{min} = 260$ ,  $\lambda_{242}/\lambda_{260} = 1.38$ ,  $\lambda_{276}/\lambda_{260} = 1.19$ , HPLC  $\lambda_{ret} = 10.0$  min (a), FD MS m/z 349.0 (M<sup>+</sup>·+1).

### 5-[R-methoxycarbonyl(hydroxy)methyl]-2-thiouridine (4b)

The persilylated ester 10b (0.27 mmol, 196 mg) was transformed to 4b according to the procedure described for 2a. 4b (88 mg, 94% yield):  $R_f$  = 0.27 (C),  $R_f$  = 0.38 (F),  $^1\text{H}$  NMR CD\_3OD (6, ppm): 8.67 (br s, 1H, H-6), 6.78 (d,  $J_{1\cdot,2}$  = 2.4 Hz, 1H, H-1'), 5.16 (s, 1H, H-C-5¹), 4.40 (dd,  $J_{1\cdot,2}$  = 2.4 Hz,  $J_{2\cdot,3}$  = 4.8 Hz, 1H, H-2'), 4.33 (dd,  $J_{3\cdot,2}$  = 4.8 Hz,  $J_{3\cdot,4}$  = 7.1 Hz, 1H, H-3'), 4.26 (dt,  $J_{3\cdot,4}$  = 7.1 Hz,  $J_{4\cdot,5}$  = 2.4 Hz, 1H, H-4'), 4.16 (dd,  $J_{4\cdot,5}$  = 2.3 Hz,  $J_{5\cdot,5}$  = 12.4 Hz, 1H, H-5'), 3.98 (dd,  $J_{4\cdot,5}$  = 2.4 Hz,  $J_{5\cdot,5}$  = 12.4 Hz, 1H, H-5'), 3.93 (s, 3H, CH<sub>3</sub>), UV (nm) pH 7 (H<sub>2</sub>O):  $\lambda_{max}$  = 280 (E = 9200), pH 12 (NaOH):  $\lambda_{max}$  = 243,  $\lambda_{max}$  = 276,  $\lambda_{min}$  = 260,  $\lambda_{243}/\lambda_{260}$  = 1.36,  $\lambda_{276}/\lambda_{260}$  = 1.20, HPLC  $T_{ret}$  = 16.5 min (a), FD MS m/z 349.0 (M<sup>+</sup>·+1).

### 5-[S-carboxy(hydroxy)methyl]uridine (1)

Nucleoside 2a (0.06 mmol, 20 mg) was dissolved in 0.01N NaOH/H<sub>2</sub>O (10 mL) and kept 48 h at room temperature. The reaction mixture was acidified with 0.1 N HCl to pH 2 and concentrated under reduced pressure. The residue was purified by paper chromatography (Whatman 3MM, solvent system E) to give 12 mg of 1 (60% yield) as a foam. R<sub>f</sub> = 0.18 (E), 1H NMR CD<sub>3</sub>OD ( $\delta$ , ppm): 8.13 (s, 1H, H-6), 5.92 (d.

 $J_{1',2'}=4.5$  Hz, 1H, H-1'), 5.02 (H-C-5<sup>1</sup> signal covered by H<sub>2</sub>O signal), 4.22-4.14 (m, 2H, H-2',3'), 4.02 (dt,  $J_{4',5'}=2.8$  Hz,  $J_{3',4'}=4.3$  Hz, 1H, H-4'), 3.85 (dd,  $J_{4',5''}=2.8$  Hz,  $J_{5',5''}=12.2$  Hz, 1H, H-5'), 3.74 (dd,  $J_{4',5''}=3.2$  Hz,  $J_{5',5''}=12.2$  Hz, 1H, H-5''). UV (nm) pH 7 (H<sub>2</sub>O):  $\lambda_{max}=260$  ( $\epsilon=6000$ ),  $\lambda_{min}=236$  ( $\epsilon=850$ ), HPLC  $T_{ret}=6.62$  min (c).

# 5-[S-carboxy(hydroxy)methyl]-2-thiouridine (3)

The nucleoside  $\underline{4a}$  (0.057 mmol, 20 mg) was transformed to  $\underline{3}$  according to the procedure described for the preparation of  $\underline{1}$ . 15 mg (79% yield);  $R_f=0.19$  (E),  ${}^1H$  NMR CD<sub>3</sub>OD ( $\delta$ , ppm): 8.48 (s, 1H, H-6), 6.60 (d,  $J_{1',2'}=2.5$  Hz 1H, H-1'), 5.00 (H-C-5<sup>1</sup> signal covered by H<sub>2</sub>O signal), 4.22 (dd,  $J_{1',2'}=2.5$  Hz,  $J_{2',3'}=4.8$  Hz, 1H, H-2'), 4.14 (dd,  $J_{2',3'}=4.8$  Hz,  $J_{3',4'}=7.0$  Hz, 1H, H-3'), 4.07 (dt,  $J_{3',4'}=7.0$  Hz,  $J_{4',5''}=2.2$  Hz, 1H, H-4'), 3.96 (dd,  $J_{4',5''}=2.2$  Hz,  $J_{5',5''}=12.4$  Hz, 1H, H-5'), 3.80 (dd,  $J_{4',5''}=2.6$  Hz,  $J_{5',5''}=12.4$  Hz, 1H, H-5''). UV (nm) pH 7 (H<sub>2</sub>O):  $\lambda_{\text{max}}=221$  ( $\in$  =7300)  $\lambda_{\text{max}}=278$  ( $\in$  =7600),  $\lambda_{\text{min}}=246$  ( $\in$  =2840), pH 12 (NaOH):  $\lambda_{\text{max}}=242$ ,  $\lambda_{\text{max}}=273$ ,  $\lambda_{\text{min}}=260$ ,  $\lambda_{242}/\lambda_{260}=1.28$   $\lambda_{273}/\lambda_{260}=1.11$ , HPLC  $T_{\text{ret}}=8.01$  min (b).

# Transformation of 4a to 2a and 4b to 2b

To  $\underline{4a}$  (3.6 mg) dissolved in phosphate buffer (0.05 M sodium phosphate, pH 8.0, 191  $\mu$ L) H<sub>2</sub>O<sub>2</sub> (30% solution, 2.27  $\mu$ L) was added. The reaction was completed after 20 h. The crude reaction mixture was co-chromatographed (HPLC) with authentic samples of  $\underline{2a}$ ,  $\underline{2b}$  and  $\underline{1}$ . The ester  $\underline{2a}$  was identified as the sole product of the S<sup>2</sup>  $\longrightarrow$  O<sup>2</sup> transformation at the substrate C<sub>2</sub> atom. An analogous reaction of  $\underline{4b}$  leads to the formation of  $\underline{2b}$  as a sole product of the substrate C<sub>(2)</sub>S  $\longrightarrow$  C<sub>(2)</sub>O transformation.

# Acknowledgement

We thank Prof. M. Kiełczewski, Department of Chemistry, University of Poznań, for valuable discussion of CD and ORD spectra.

This investigation was supported by CPBP 3.13 and RP II 10 grants.

### REFERENCES

- 1. Małkiewicz, A.J.; Nawrot, B.; Sochacka, E. Z.Naturforsch. 1987, 42b, 360.
- Kawakami, M.; Tsonis, P.A.; Nishio, K.; Takemura, S. J. Biochem. 1980, 88, 1151.
- 3. Garel, J.P.; Keith, G. Nature. 1977, 269, 350.
- Kawakami, M.; Nishio, K.; Takemura, S.; Kondo, T.; Goto, T. Nucleic Acids Symp. Ser. 1979, No. 6, s53.
- Kondo, T.; Fukami, T.; Goto, T.; Kawakami, M.; Takemura,
   Nucleic Acids Symp. Ser. 1983, No. 12, 127.
- 6. Garel, J.P. Nature, 1976, 260, 805.
- Smith, D.W.E.; Mc Namara, L.; Rice, M.; Hatfield, D.C. J.Biol.Chem. 1981, 256, 10033.
- B. Buckingham, R.H., Grosjean, H. In: Galas, P.J.; Kirkwood, B.; Rosenberg, F.F. (eds.) Accuracy in molecular processes: the control and reliance to living systems. Chapman and Hall Ltd., London, 1985, 83-126.
- Adamiak, R.; Górnicki, P. Prog. Nucl. Ac. Res. Mol. Biol. 1985, 32, 27.
- Yokoyama, S.; Yamaizumi, Z.; Nishimura, S.; Miyazawa, T. Nucl. Acids Res. 1979, 8, 2611.
- 11. Yokoyama, S.; Watanabe, T.; Murao, K.; Ishikura, H.; Yamaizumi, Z.; Nishimura, S.; Miyazawa, T. Proc.Natl. Acad.Sci. USA, 1985, 82, 4905.
- 12. Kawai, G.; Hara-Yokoyama, M.; Miyazawa, T.; Hasegawa, T.; Murao, K.; Ishikura, H.; Nishimura, S.; Yokoyama, S. 12th International tRNA Workshop, Umea, Sweden, 1987, IIIa-15.
- 13. Hillen, W.; Egert, E.; Lindner, H.J.; Gassen, H.G. Biochemistry, 1978, 17, 5314.
- Sierzputowska-Gracz, H.; Sochacka, E.; Małkiewicz, A.; Kuo, K.; Gehrke, Ch.W.; Agris, P.F. J. Am. Chem. Soc. 1987, 109 7171.
- 15. Małkiewicz, A.; Nawrot, B.; Sochacka, E. Nucleic Acids Symp. Ser. 1987, No. 18, 97.
- Hayakawa, H.; Tanaka, H.; Obi, K.; Itoh, M.; Miyasaka, T. Tetrahedron Lett. 1987, 28, 87.
- Ogilvie, K.K.; Schifman, A.L.; Penney, Ch.L. Can. J. Chem. 1979, 57, 2230.
- 18. Wolf, F.J.; Weyland, J.; Leonard, N.Y.; Miller, L.A. Organic Synthesis Coll. 1963, Vol. 4, 124: n-butyl glyoxylate was distilled under reduced pressure over P<sub>2</sub>O<sub>5</sub> and in a stream of argon just before the condensation.
- 19. Korver, O. Tetrahedron, 1970, 26, 5507.
- 20. Barth, G.; Voelter, W.; Mosher, H.S.; Bunnenberg, E.; Djerassi, C. J. Am. Chem. Soc. 1970, 92, 875.
- 21. Ueda, T.; Nishino, H. Chem. Pharm. Bull. 1969, 17, 920.

- 22. Watanabe, K. Biochemistry, 1980, 19, 5542.
- 23. Under described conditions ester 2a hydrolyzes exclusively to acid  $1 (T_{Ret} = 6.62 \, \text{min}^C)$ , whereas ester 2b gaves only  $5-[R-carboxy(hydroxy)methyl]uridine <math>(T_{Ret} = 6.02 \, \text{min}^C)$ .

Received March 3, 1989.